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Atoms CHAPTER 4 REVIEW

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Answers Chapter 4 SECTION 1

SHORT ANSWER 1. In order for

an electron to be ejected

from a metal surface, the

electron must be struck by a

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**In Atoms** single photon with at least the minimum energy needed to knock the electron loose.

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Holt Modern Chemistry Review

CHAPTER 4: ARRANGEMENT OF  
ELECTRONS IN ATOMS. The

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bulk (but not all) of the  
information for the chapter  
4 test. Focus on this  
content, but make sure to



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review class notes,  
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questions, etc.

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Transcription. 1 CHAPTER 4  
PRE-TEST Arrangement of

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**In Atoms** in Atoms In the space provided, write the letter of the term that best completes each 3 Chapter 4 PRE TEST, continued 8. What is the frequency of a photon whose energy is J? ( $h = \text{J s}$ )  
a Hz b Hz c Hz d Hz 9. When

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of Electrons in Atoms

SECTION 2 SHORT ANSWER

Answer the following questions in the space provided. 1. d How many quantum numbers are used to describe the properties of electrons in atomic

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orbitals? (a) 1 (c) 3 (b) 2

(d) 4 2. a A spherical electron cloud surrounding an atomic nucleus would best represent (a) an s orbital.

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Modern Chemistry 28 Chapter  
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Test A, continued \_\_\_\_\_ 8.

What is the frequency of a  
photon whose energy is  $3.4 \times 10^{-19}$  J? ( $h = 6.626 \times 10^{-34}$  J·s)

a.  $8.8 \times 10^{26}$  Hz   b.  $5.1 \times 10^{14}$   
Hz   c.  $1.9 \times 10^{-15}$  Hz   d. 2.3

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10-52 Hz \_\_\_\_\_ 9. When electromagnetic radiation strikes the surface of a metal, electrons are ejected from the metal's ...



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Atoms . SHORT ANSWER Answer  
the following questions in  
the space provided. 1.

\_d=--\_ How many quantum

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**In Atoms** are used to describe the properties of electrons in atomic orbitals? (a) 1 (c) 3 (b) 2 (d) 4 . 2.

\_a,--\_ A spherical electron cloud surrounding an atomic nucleus would best represent

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Key Concepts Chapter 4 test  
arrangement of electrons in  
atoms. The attractions  
between the protons and  
electrons of atoms can cause  
an electron to move  
completely from one atom to

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the other. When an atom loses or gains an electron, it is called an ion Chapter 4 test arrangement of electrons in atoms.

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ELECTRONS IN ATOMS 93 FIGURE  
4-3 The photoelectric  
effect: electromagnetic  
radiation strikes the  
surface of the metal,

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**In Atoms** ejecting electrons from the metal and creating an electric current. Frequency and wavelength are mathematically related to each other.

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